ATP (adenosine triphosphate) is the cell's main energy source in energy coupling
ATP = adenine + ribose + 3 phosphates



When the bonds between the phosphate groups are broken by hydrolysis → Energy is released
 This release of energy comes from the chemical change to a state of lower free energy, not in the phosphate bonds themselves



HOW ATP PERFORMS WORK

- *Exergonic* release of P_i is used to do the *endergonic* work of cell
- When ATP is hydrolyzed, it becomes ADP (adenosine diphosphate)



• <u>Catalyst</u>: substance that can change the rate of a reaction without being altered in the process nzyme + biological catalyst L> speeds up the reaction but the end product does not change. Sucrase GlucoseFructose $C_6H_{12}O_6$ $(C_6H_{12}O_6)$ Sucrose 0 $(C_{12}H_{22}O_{11})$ breaking bonds) lipase - cuts lipide anglase -> cuts anylose phospholipase



Progress of the reaction —>

SUBSTRATE SPECIFICITY OF ENZYMES

- The reactant that an enzyme acts on is called the enzyme's substrate
- The enzyme binds to its substrate, forming an enzyme-substrate complex
- The active site is the region on the enzyme where the substrate binds





2 Substrates are held in active site by weak interactions.



Enzyme-substrate complex



Substrates

2 Substrates are held in active site by weak interactions.



Substrates are converted to products.









Step 1

Step 2

Step 3





INDUCED FIT: ENZYME FITS SNUGLY AROUND SUBSTRATE -- "CLASPING HANDSHAKE"



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An enzyme's activity can be affected by:

- Temperature
- pH∨
- Chemicals

denatured



ENZYME STRUCTURE & FUNCTION

- Change to the molecular structure of a component in an enzymatic system may result in a change of function or efficiency of the system
- **Denaturation**: disrupt protein structure
 - \rightarrow reduce enzymatic activity
- **Environmental pH**: alter efficiency of enzyme activity; disruption of H-bonds
- In some cases, enzyme denaturation is *reversible* → enzyme regains activity

COFACTORS

- <u>Cofactors</u>: nonprotein enzyme helpers such as minerals (eg. Zn, Fe, Cu)
- <u>Coenzymes</u>: organic cofactors (eg. vitamins)

Enzyme Inhibitors

- <u>Competitive inhibitor</u>: binds to the *active site* of an enzyme, competes with substrate
- Noncompetitive inhibitor: binds to another part of an enzyme → enzyme changes shape → active site is nonfunctional

ENZYME SPECIFICITY



Figure 1: Enzymesubstrate complex

Figure 2: The charges align between the enzyme and the substrate; however, the enzyme's shape will not "fit". Figure 3: The shape of the substrate appears to fit but the charges do not align in the active site of the enzyme.



NONCOMPETITIVE INHIBITION Attaches at a site other them the active site



INHIBITION OF ENZYME ACTIVITY



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REGULATION OF ENZYME ACTIVITY

- To regulate metabolic pathways, the cell switches on/off the genes that encode specific enzymes
- <u>Allosteric regulation</u>: protein's function at one site is affected by binding of a regulatory molecule to a separate site (allosteric site)
 - Activator stabilizes active site
 - Inhibitor stabilizes <u>inactive</u> form
 - Cooperativity one substrate triggers shape change in other active sites → increase catalytic activity



(b) Cooperativity: another type of allosteric activation

Substrate



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Stabilized active form

FEEDBACK INHIBITION

• End product of a metabolic pathway shuts down pathway by binding to the allosteric site of an enzyme

• Prevent wasting chemical resources, increase efficiency of cell



ORGANIZATION OF ENZYMES WITHIN A CELL



The matrix contains enzymes in solution that are involved in one stage of cellular respiration.

> Enzymes for another stage of cellular respiration are embedded in the inner membrane.

